

PAHs and PAH clusters in the far-IR

Alexander K. Lemmens^{1,2}, Sébastien Gruet^{3,4,5}, Amanda L. Steber^{3,4,5}, Jens Antony⁶, Stefan Grimme⁶, Melanie Schnell^{3,4,5}, Anouk M. Rijs¹

¹Radboud University, Institute for Molecules and Materials, FELIX Laboratory,
Toernooiveld 7c, 6525 ED Nijmegen, The Netherlands. E-mail: a.rijs@science.ru.nl

²Van't Hoff Institute for Molecular Sciences, University of Amsterdam,
Science Park 904, 1098 XH Amsterdam, The Netherlands

³Deutsches Elektronen-Synchrotron, Notkestrasse 85, D-22607 Hamburg, Germany.
E-mail: amanda.steber@desy.de

⁴Institut für Physikalische Chemie, Christian-Albrechts-Universität zu Kiel,
Max-Eyth-Strasse 1, D-24118 Kiel, Germany

⁵The Hamburg Centre for Ultrafast Imaging, Luruper Chaussee 149,
D-22761 Hamburg, Germany

⁶Mulliken Center for Theoretical Chemistry, University of Bonn, Beringstr. 4,
D-53115 Bonn, Germany

Polycyclic aromatic hydrocarbon (PAH)-water complexes are an important link between molecular gas and condensed particles in both the earth's atmosphere and the interstellar medium¹. We report on the experimental and theoretical investigations of the stepwise complexation of the PAH acenaphthene with itself and with water². The far-IR region of the IR spectrum is especially interesting since it reveals spectral features that result from the complexation and/or microhydration and is therefore very sensitive for the molecular structure. Using mass- and conformer selective far-IR action spectroscopy, we show that the water clusters are locked with little perturbation on the PAH 'platform'³. Density functional theory (DFT) calculations to analyse our observed signatures are still challenging in predicting the far-IR frequency range, especially when hydrogen bond interactions are involved, although applying anharmonic corrections leads to improvements.

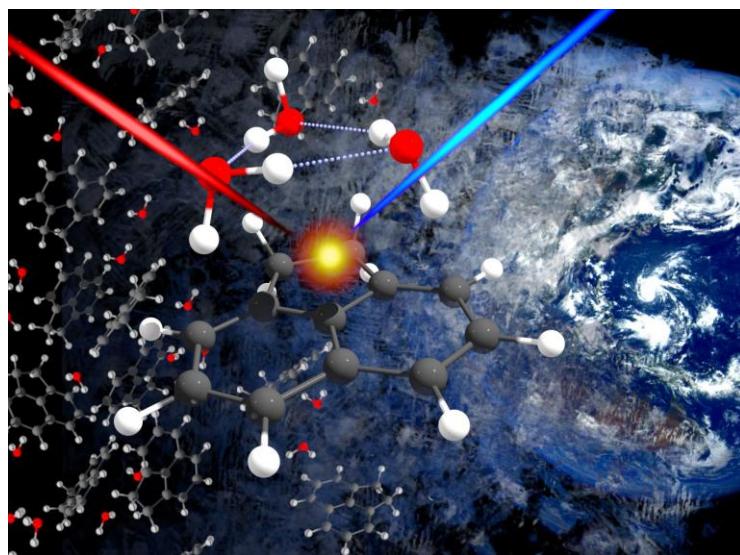


Figure 1: Impression of PAHs and PAH clusters studied in the far-IR, in both the earth's atmosphere and interstellar space

References

- [1] A. Potapov, *Mol. Astrophys.*, 2017, 6, 16–21.
- [2] A. K. Lemmens, S. Gruet, A. L. Steber, J. Antony, S. Grimme, M. Schnell, A. M. Rijs, *Phys. Chem. Chem. Phys.*, 2019, Advance Article
- [3] A. L. Steber, C. Pérez, B. Temelso, G. C. Shields, A. M. Rijs, B. H. Pate, Z. Kisiel and M. Schnell, *J. Phys. Chem. Lett.*, 2017, 8, 5744–5750.